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1. A is a solution of hydrochloric acid. B is a solution containing 2.45 g of anhydrous sodium trioxocarbonate (IV) in 250 cm³ of solution. (a) Put A into the burette and titrate it against 20.0 cm³ or 25.0cm³ portion of B using methyl orange as indicator.Repeat the exercise to obtain consistent titres.

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